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The chapter additionally differentiates between transition elements and main elements, elaborating on their features and characteristics. It also describes the idea of energy of ionization and its trends among groupings and periodic groups on the table of elements¹. The chapter talks about atomic and ionic, and how these can be predicted based on an atom's place on the periodic table.

The concept of electronegativity, which calculates an atom's propensity to attract a bonding pair of electrons, is another crucial subject covered in this section¹. The chapter ends with a dialogue on the diagonal relationship, character of metals, and point of boiling of components.

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