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The chapter additionally distinguishes between transition metals and main group elements, elaborating on their characteristics and traits. It also explains the concept of energy of ionization and its trends among groupings and periodic groups on the periodic table¹. The chapter discusses atom’s and radii, and how these can be forecasted based on an element’s location on the periodic table.

Electronegativity, which quantifies an atom’s tendency to pull a pair of bonding electrons, is another key theme covered in this chapter¹. The chapter concludes with a dialogue on the diagonal relationship, character of metals, and point of boiling of elements.

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The concept of electronegativity, which quantifies an atom’s propensity to draw a bonding pair of electrons, is another key theme discussed in this part¹. The chapter finishes with a conversation on the diagonal relationship, metallic character, and boiling point of elements.

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