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The chapter further differentiates between transition metals and main group elements, detailing on their features and characteristics. It also clarifies the concept of ionization energy and its trends among groupings and periodic groups on the table of elements¹. The chapter discusses atomic and ionic, and how these can be anticipated based on an atom's position on the periodic table.

Electronegativity, which measures an atom's inclination to attract a pair of bonding electrons, is another important topic addressed in this section¹. The chapter finishes with a dialogue on the relationship diagonally, metallic character, and point of boiling of components.

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